

Its More than Climate Change: The Fate of Persistent Organic Pollutants in the Arctic

Maya Wei-Haas, Kim Hageman, Cleo Davie-
Martin, Brandon McAdams, Yu-Ping Chin
1/26/17



Image of Bow Lake
from: <http://www.owallpapers.com>

Why Study Contaminants in the Arctic?



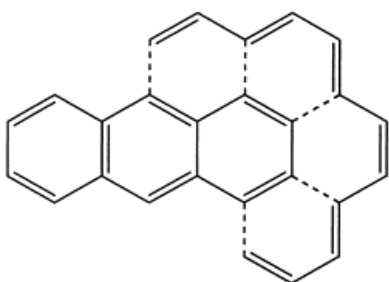
- Climate change garners the most attention
- Less known is known about the effects of persistent organic pollutants (POPs), especially in the U.S. Arctic



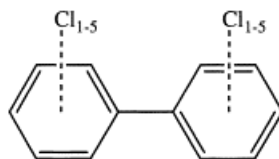
What are POPs and why do we care about them in the Arctic



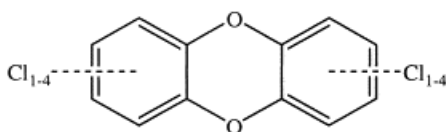
- POPs currently on the Stockholm Convention List
 - The original POPs
 - Aldrin, Chlordane, DDT, Dieldrin, Endrin, Heptachlor, HCB, Mirex, Toxophene, PCB, PCDD, PCDF
 - The new POPs (approved 5/15)
 - α , β -HCH
 - Chlordecone
 - HBB
 - HB-cyclododecane
 - PBDE
 - Hexachlorobutadiene
 - Lindane
 - PentaCB
 - PFOS, PFOA



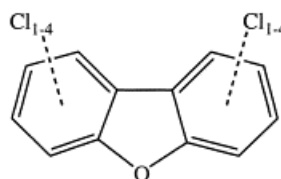
Polynuclear aromatic hydrocarbons (PAHs)



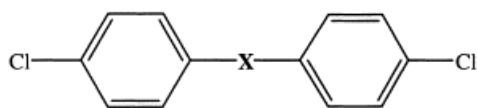
Polychlorinated biphenyls (PCBs)



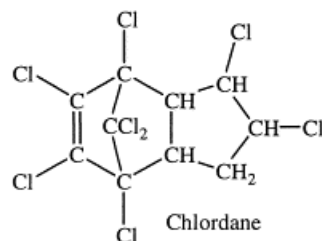
Polychlorinated dibenzodioxins (PCDDs)



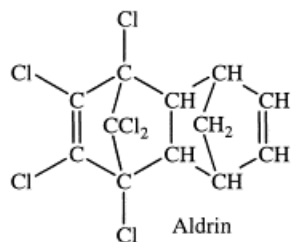
Polychlorinated dibenzofurans (PCDFs)



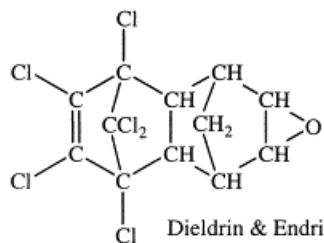
DDT, DDE, DDD ($X = \text{CH-CCl}_3$; C=CCl_2 ; CH-CHCl_2)



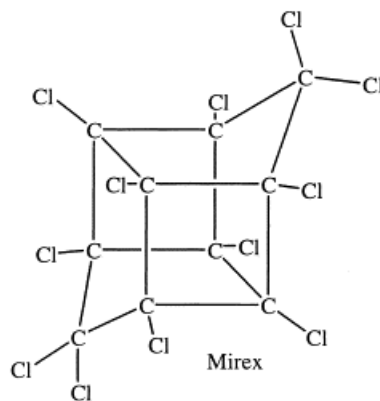
Chlordane



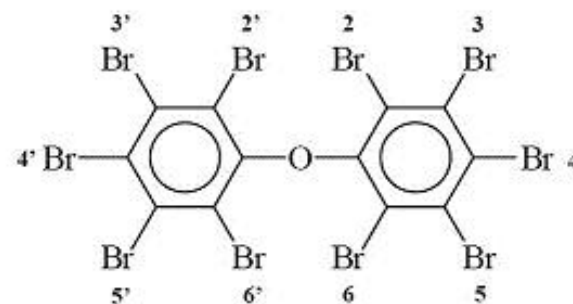
Aldrin



Dieldrin & Endrin



Mirex



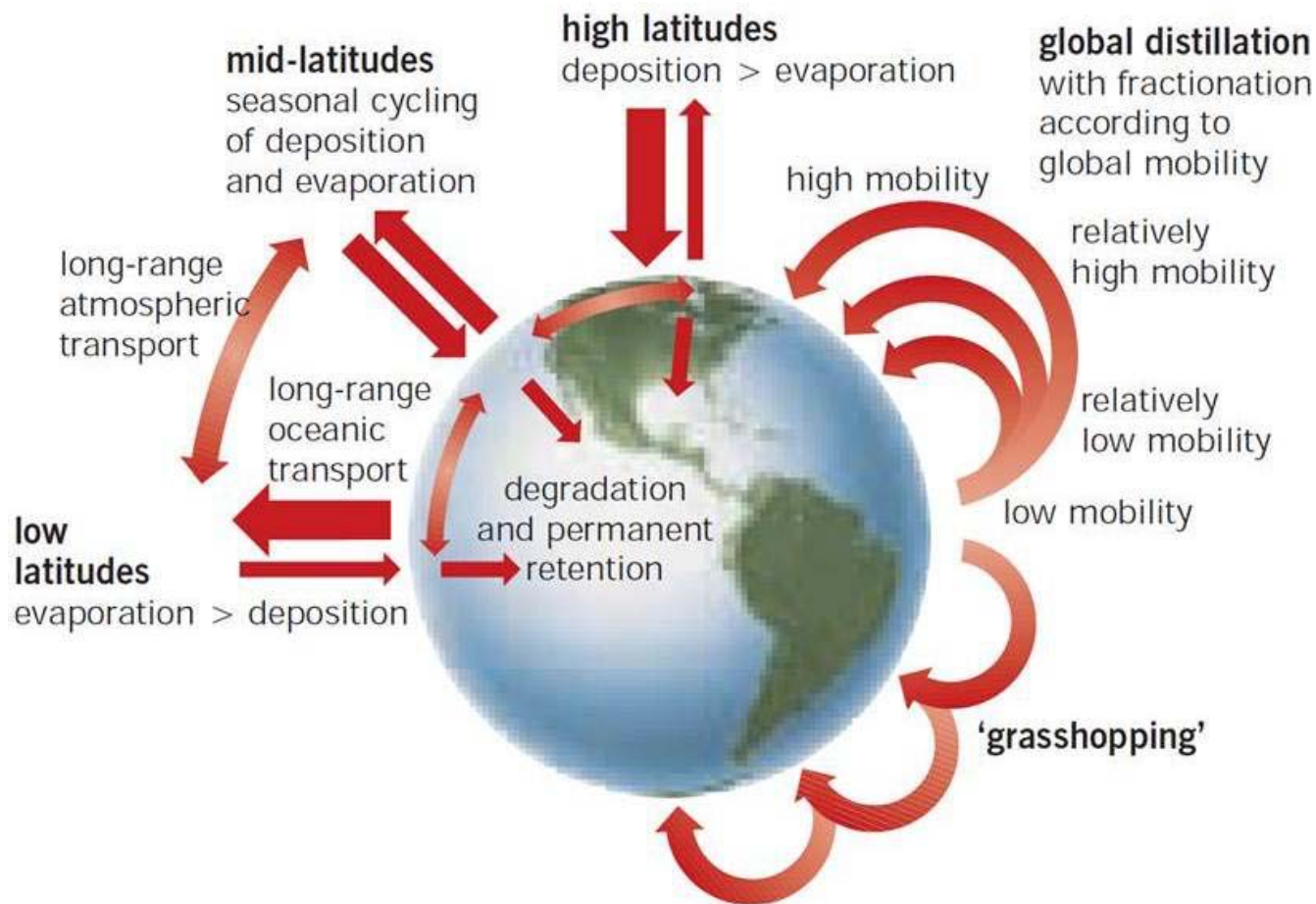
How do they get there?

- **Cold-Condensation (trapping)**
 - Contaminants sorb to water vapors (clouds/fog)
 - Deposition via precipitation
 - Snow vs. Rain
- **Global Distillation**
 - “Grasshopper effect”
 - Multiple cycles of evaporation - cold condensation
 - Transport of low latitude chemicals to the poles
- **Particle Bound Transport (dust)**
 - Particularly important for the large, hydrophobic compounds
- **Wind influences**
 - Trade forced winds
 - Diurnal influences
 - Monsoon Winds



How do POPs get to the Arctic?

Migration of persistent organic pollutants



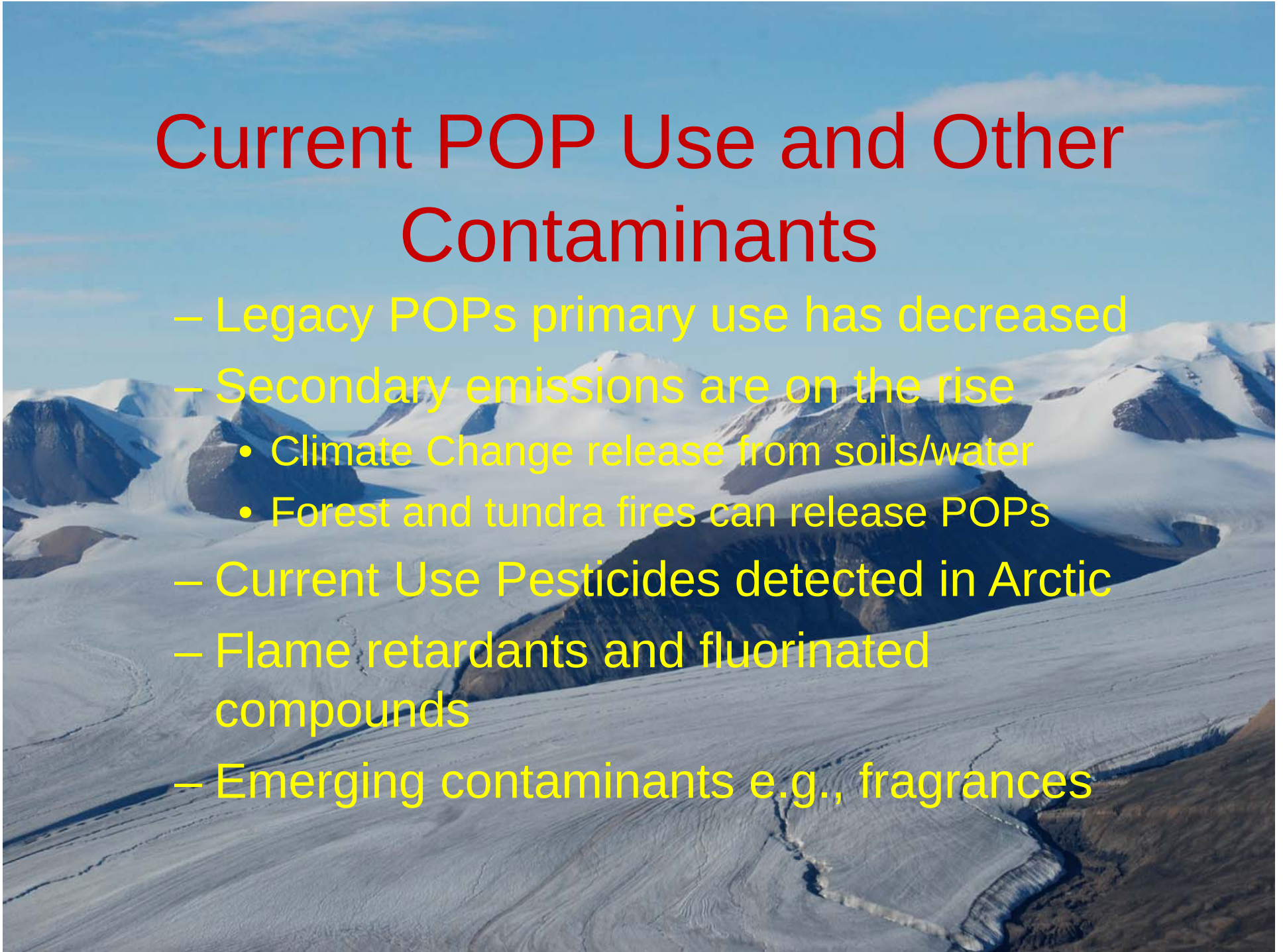
Many POPs are semi-volatile e.g., PBDE and can be mobilized through repeated volatilization/condensation steps

POPs in the Arctic

- Why are we concerned?
 - Biomagnification!!!
 - POPs are not soluble (dislike) water
 - BUT, they are soluble (like) in fat in organisms
 - Accumulate in organisms
 - Could impact organisms higher up on the food chain e.g., polar bears, wolves, people
- POPs are also very persistent
 - Do not degrade readily biotically or chemically
 - Accumulate over time

Current POP Use and Other Contaminants

- Legacy POPs primary use has decreased
- Secondary emissions are on the rise
 - Climate Change release from soils/water
 - Forest and tundra fires can release POPs
- Current Use Pesticides detected in Arctic
- Flame retardants and fluorinated compounds
- Emerging contaminants e.g., fragrances



**“Secondary” sources of
Legacy POPs:
revolatilization from
water and soil!**

Environ. Sci. Technol. **2010**, *44*, 6978–6984

Atlantic Ocean Surface Waters Buffer Declining Atmospheric Concentrations of Persistent Organic Pollutants

LUCA NIZZETTO,^{*,†,‡}
RAINER LOHMANN,^{*,§}
ROSALINDA GIOIA,[†] JORDI DACHS,^{||} AND
KEVIN C. JONES[†]

GEOPHYSICAL RESEARCH LETTERS, VOL. 36, L24602, doi:10.1029/2009GL041340, 2009

Cycling of DDT in the global environment 1950–2002: World ocean returns the pollutant

Irene Stemmler^{1,2} and Gerhard Lammel^{3,4}

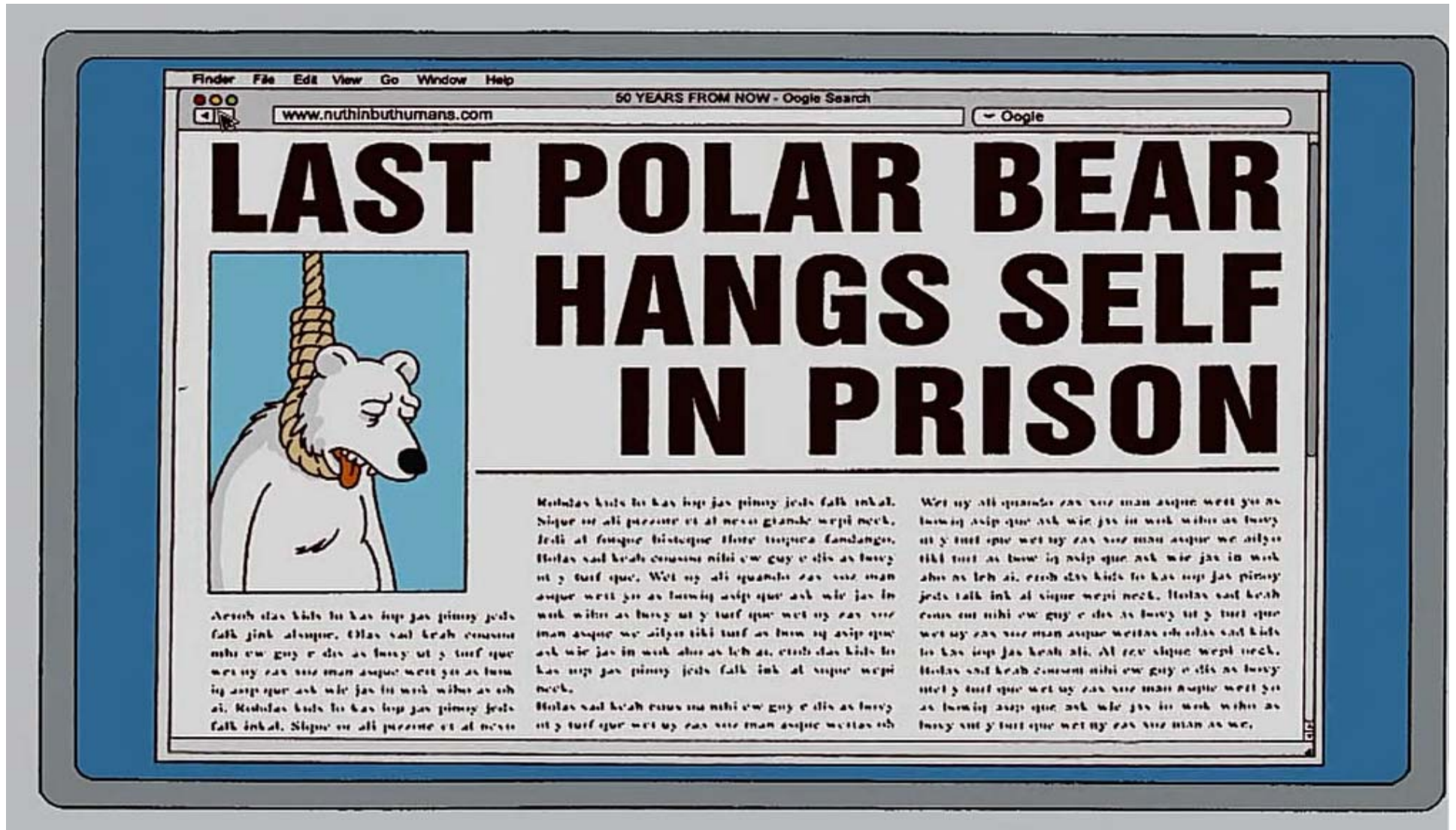
Luca Nizzetto et al.

Environ. Sci. Technol. **2010**, *44*, 6526–6531

ENVIRONMENTAL
Science & Technology
feature

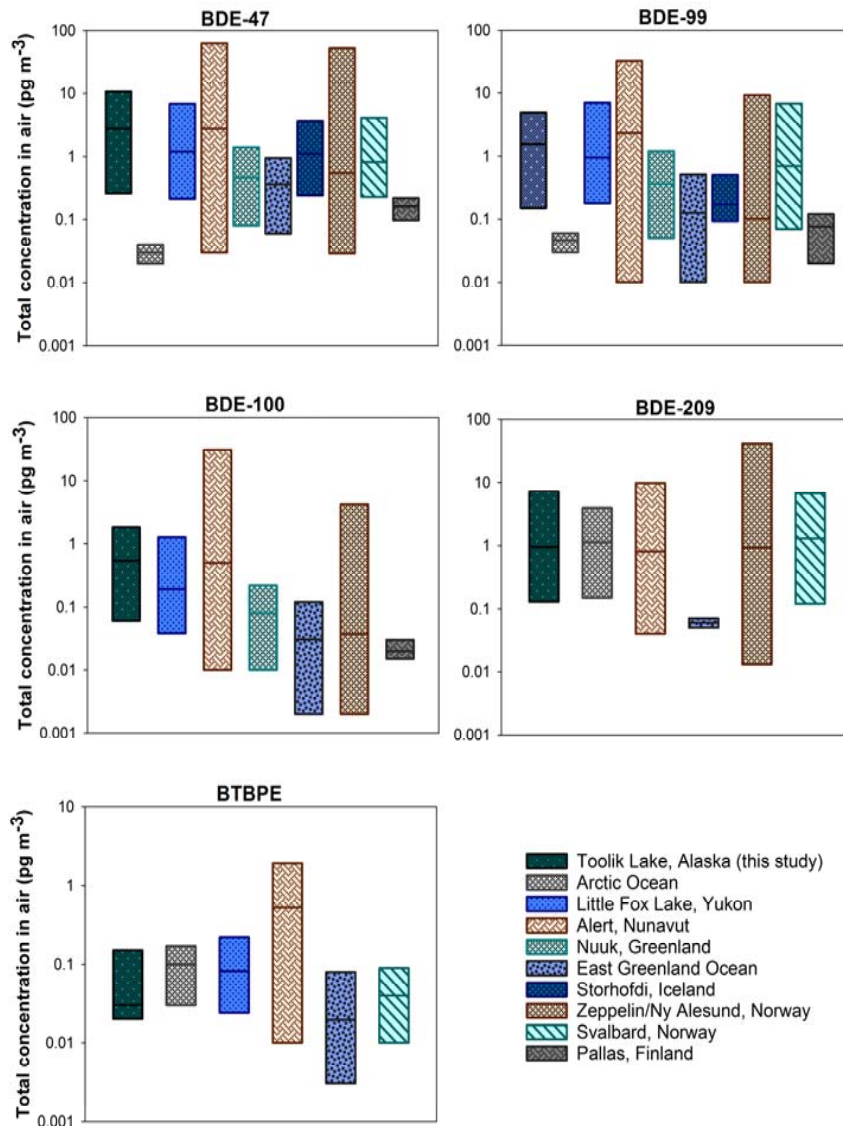
Past, Present, and Future Controls on Levels of Persistent Organic Pollutants in the Global Environment

Most POP Research in the Arctic Involve Measurements and Monitoring



What did we do at Toolik in 2011, 2013 and 2014?

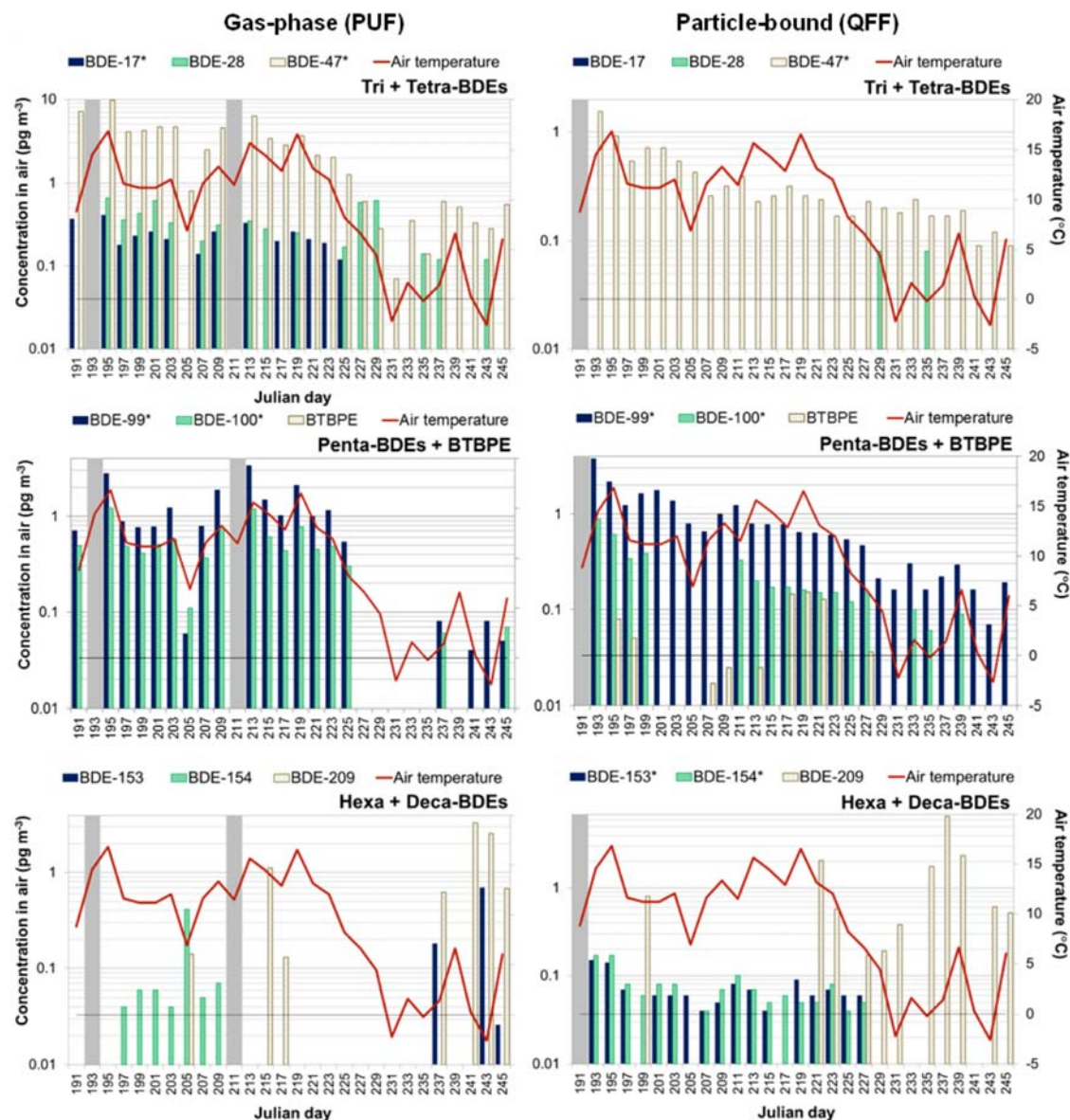
- 2011: Isolation of dissolved organic matter (DOM) samples
 - Selection and collection of surface waters
 - Large scale isolation by solid phase extraction
- 2013: More DOM isolation, PBDE air measurements, DOM binding experiments
 - Active and passive air samplers
 - PBDE binding to Arctic surface water DOM
- 2014: Actinometry and photolysis experiments



Comparison of PBDE Air Concentrations from Different Arctic Locations

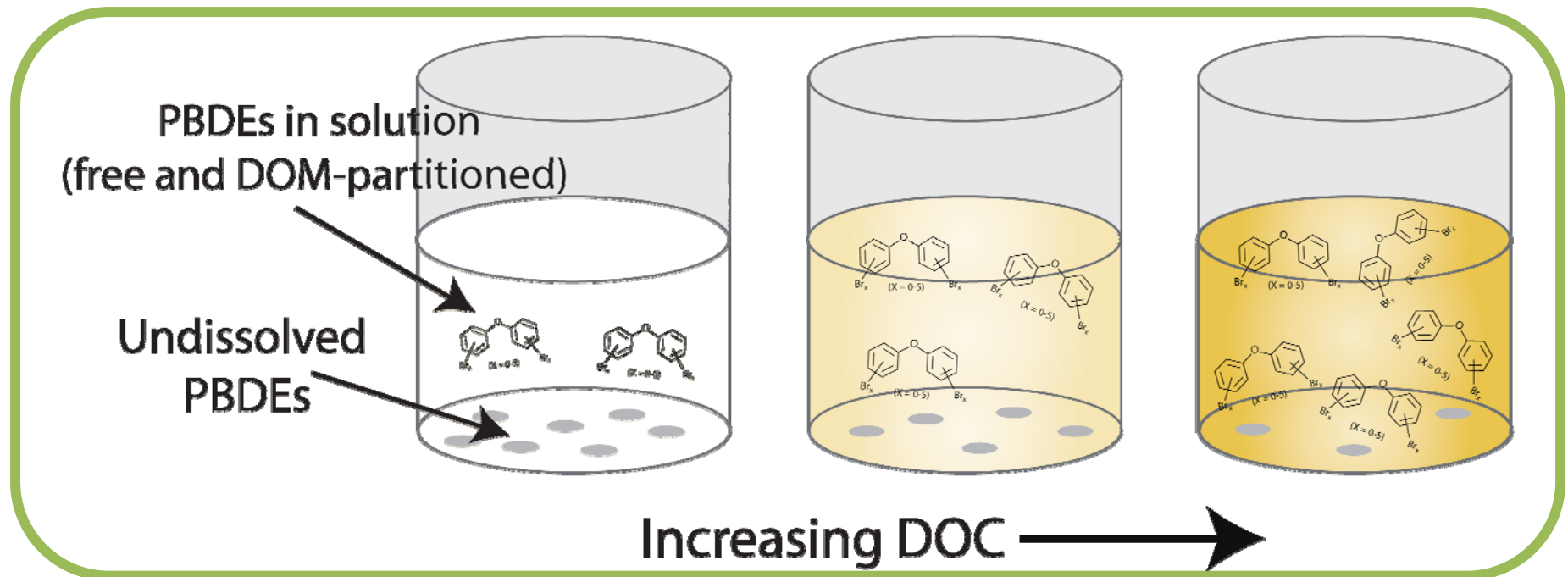
Toolik air concentrations typically ~ an order of magnitude agreement with other sites

PBDE Concentrations in Toolik Air



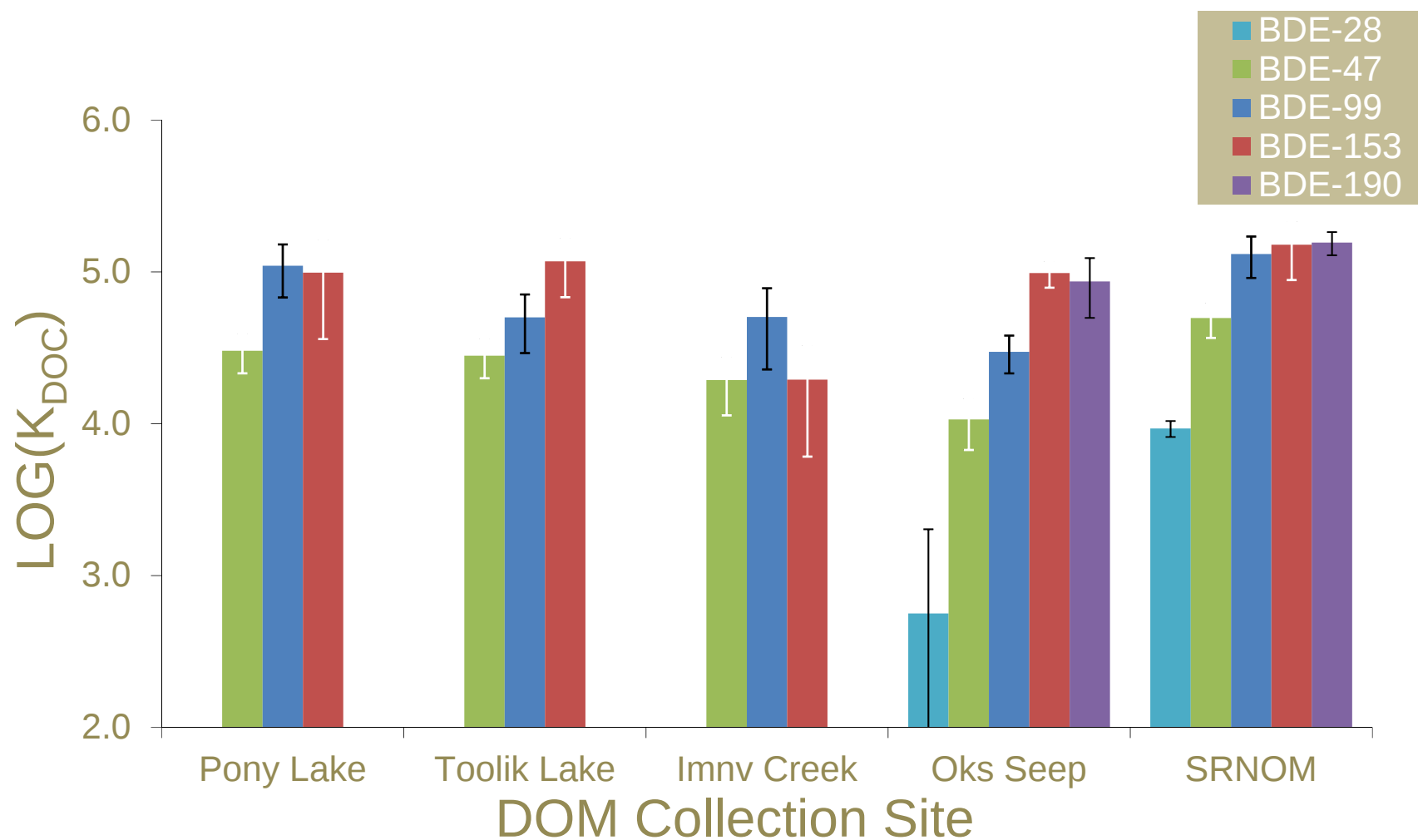
Davie-Martin et al., 2016
 Env. Sci.: Proc. Impacts,
 18, 1274-1284

Solubility Enhancement



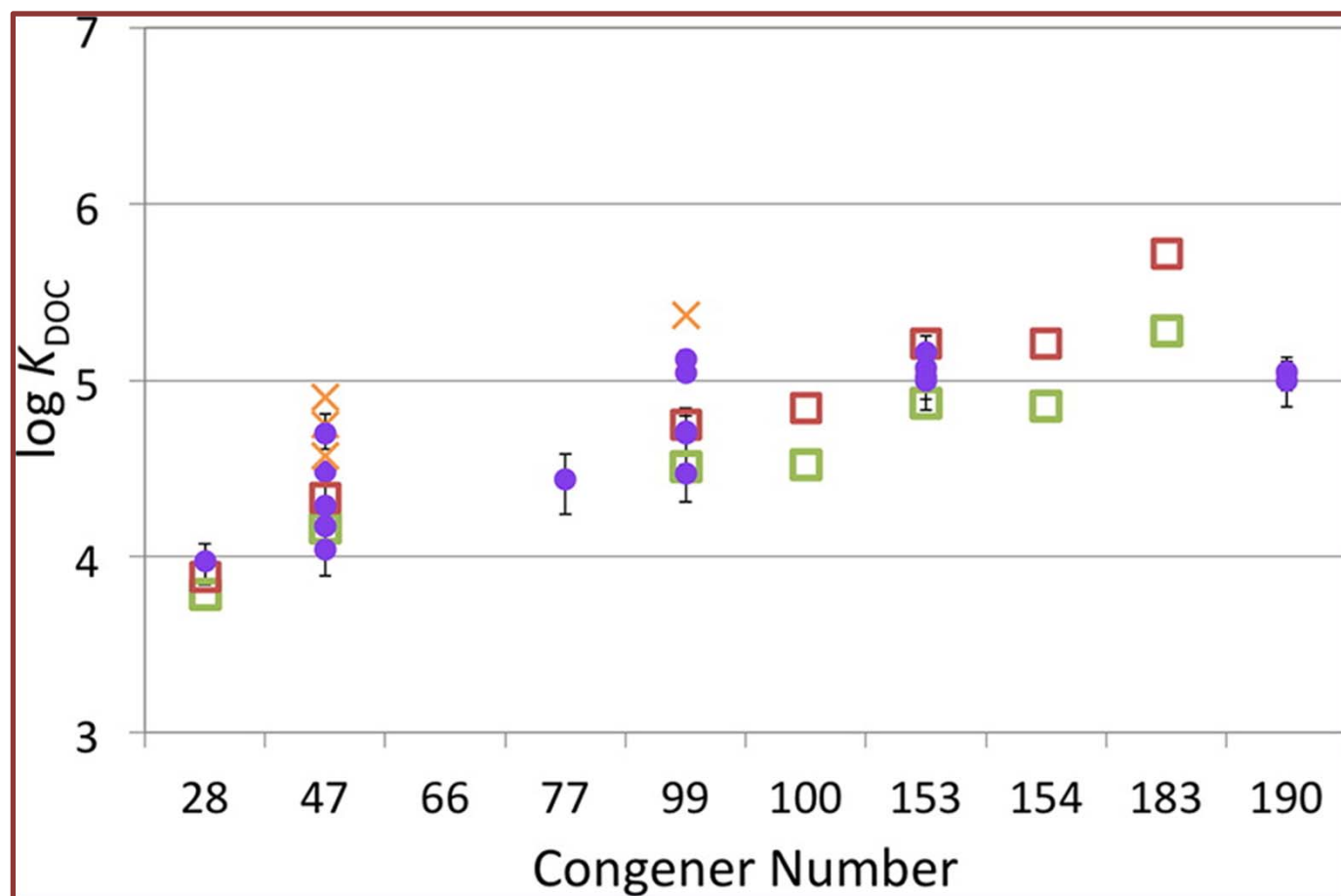
- 7 concentrations of DOM (+1 experiment with Milli-Q only)
- 4 replicates of each DOM concentration (+1 control w/o PBDE)
- 6 day minimum equilibration (kinetics study verification)
- 2:1 Liquid-liquid extraction with hexane
- GC – ECD analysis, S/SL injector, 15m RTX-1614 column

Partition Coefficient Summary

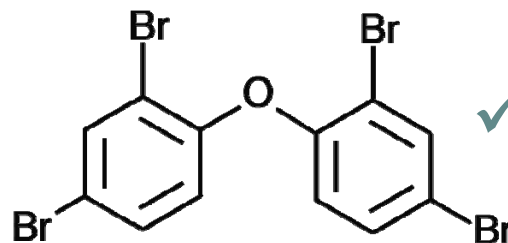
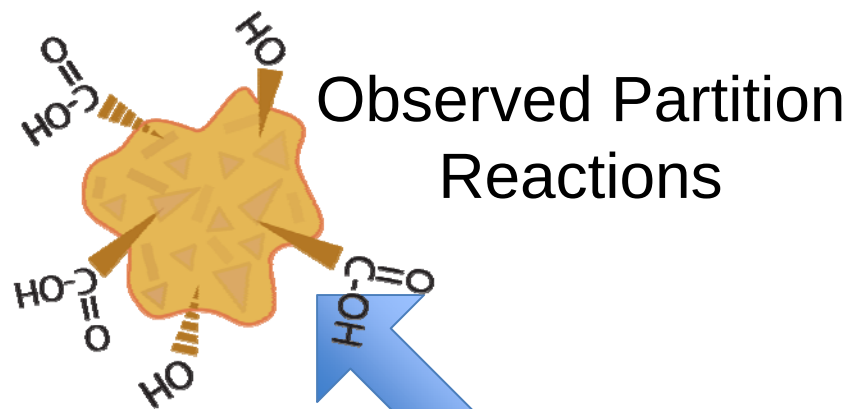


Molecular Modeling Results

ESABV = red
squares
LSABV = green
squares
Our data = filled
purple circles
Lit. Data = orange X

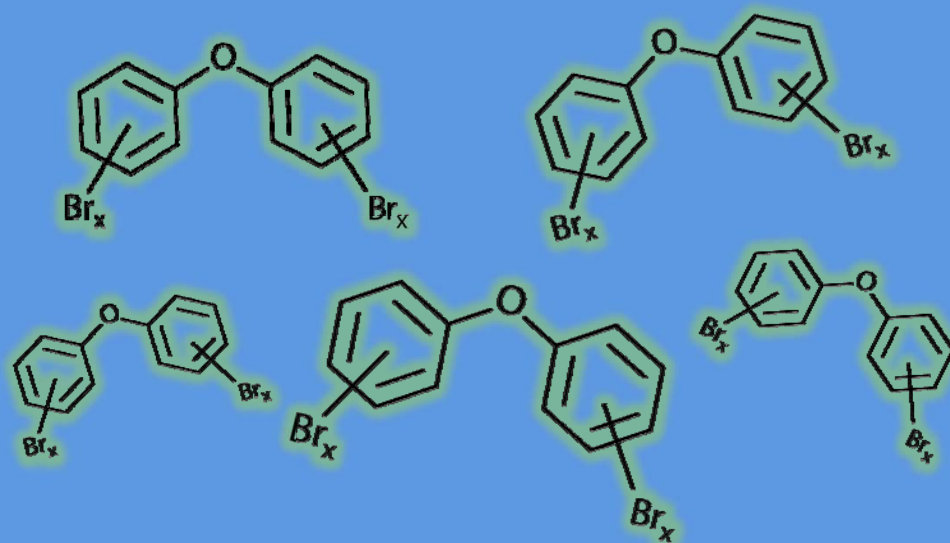
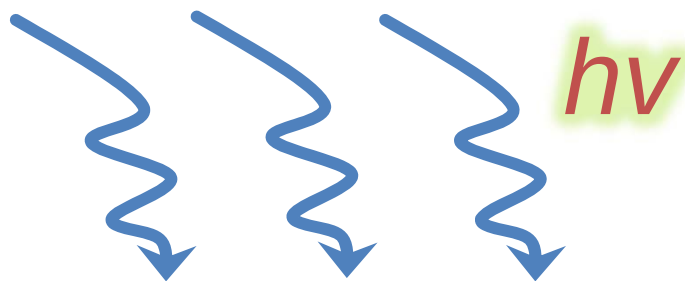


Conclusions 1:



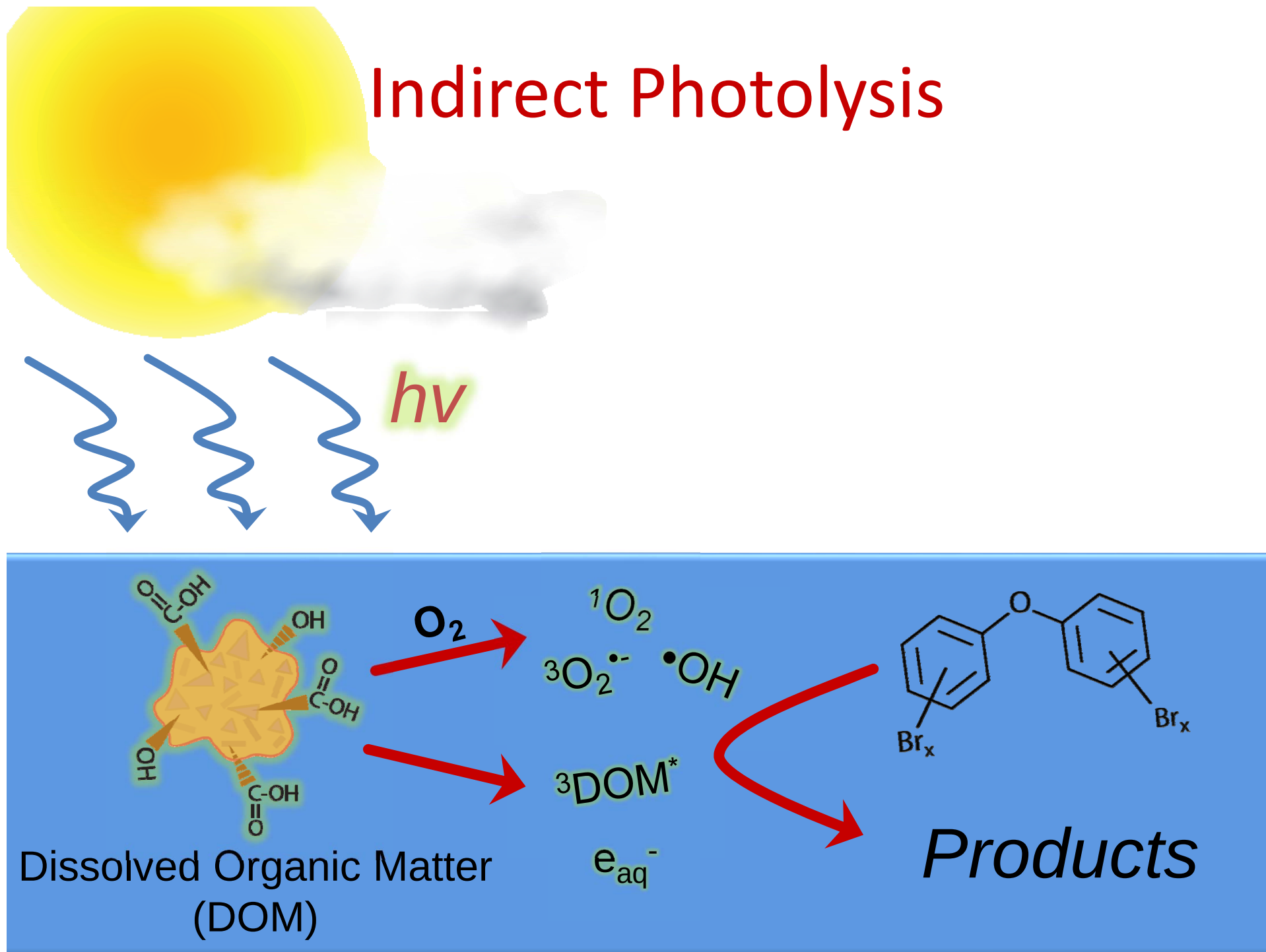
- ✓ K_{DOC} is a function of bromination and DOC properties
- ✓ pp-LFER accurately estimates this reaction
- ✓ Both ESABV and LSABV robust

Direct Photolysis



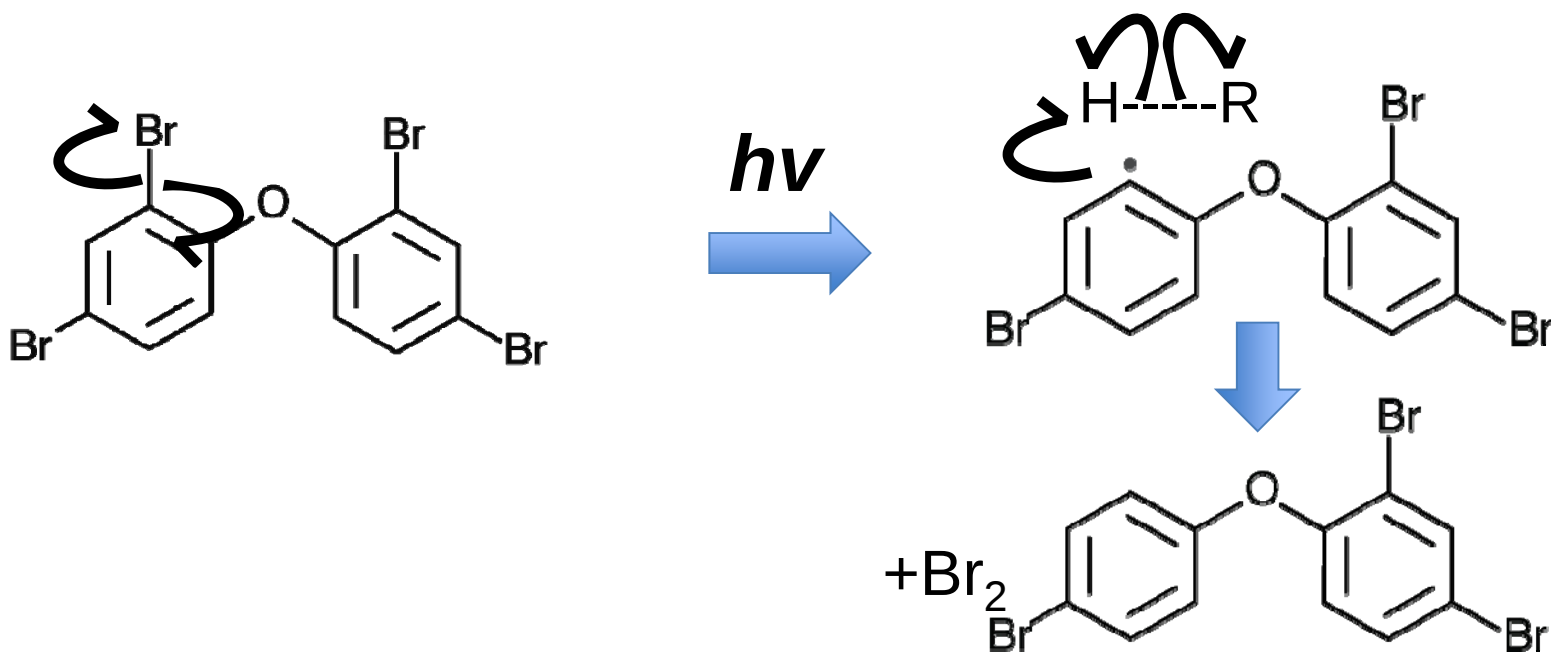
 *Products*

Indirect Photolysis

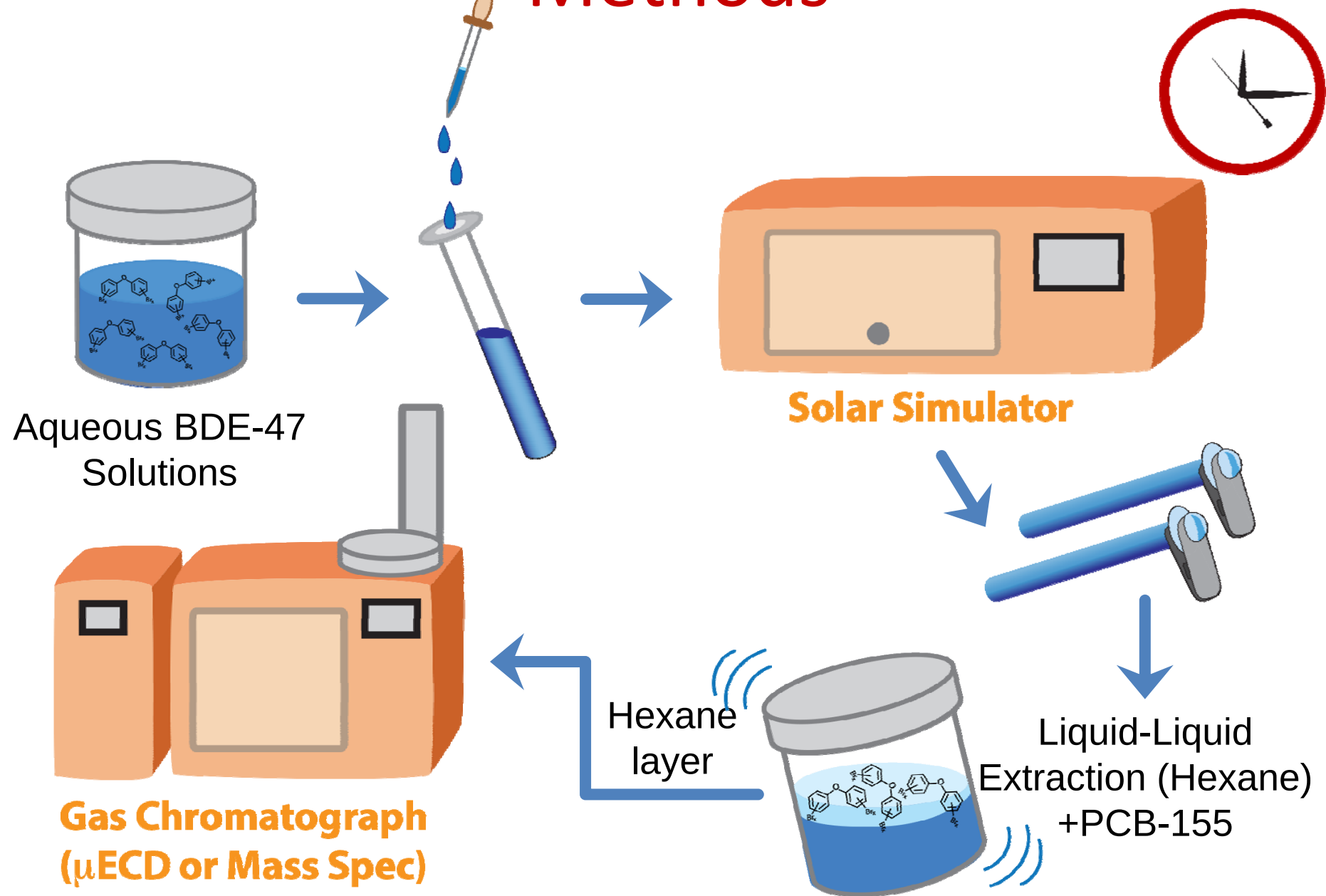


Previous Research

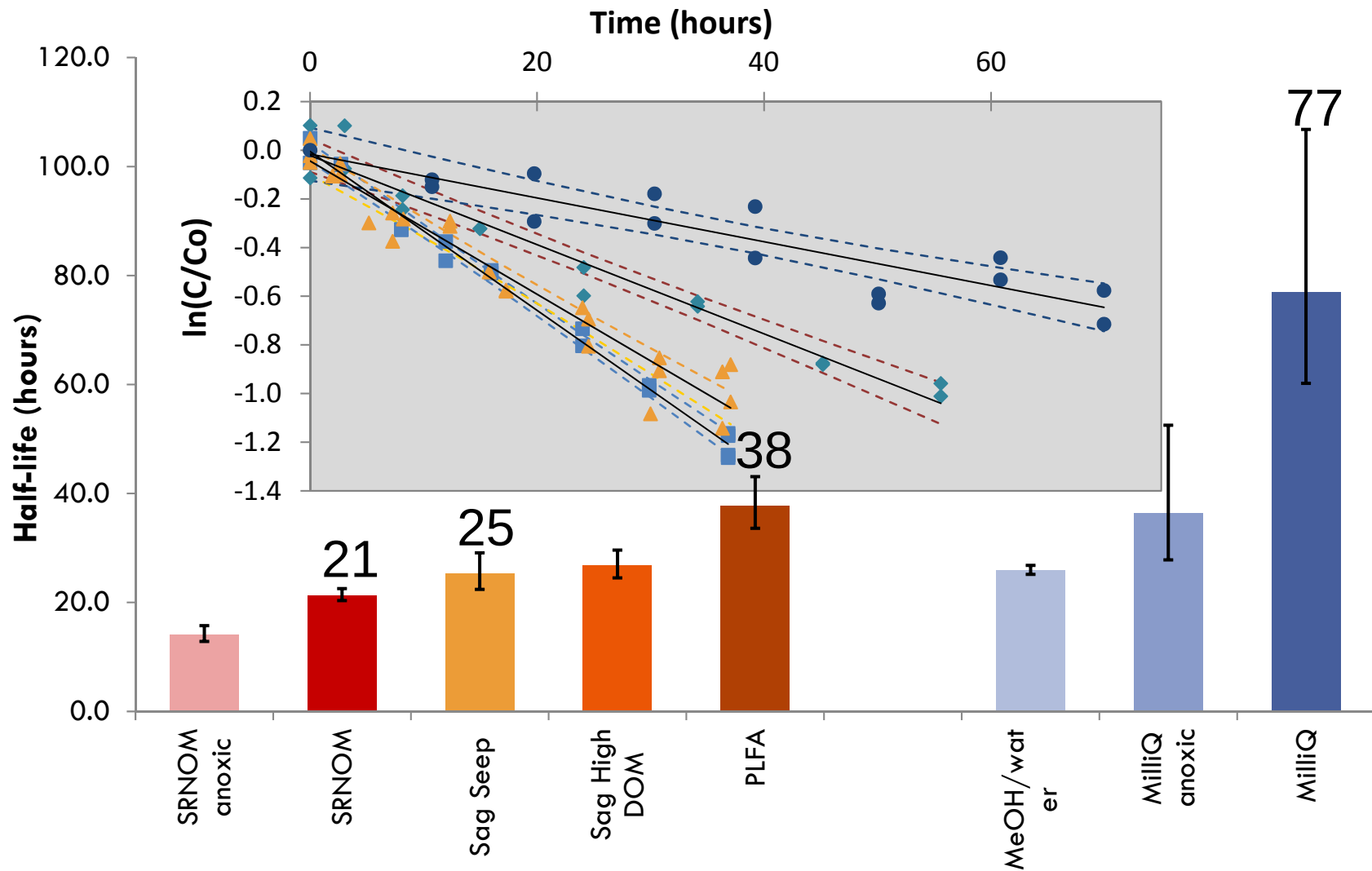
- * Direct photolysis in organic solvents
 - * $t_{1/2}$ = minutes to hours
- * Indirect photolysis and the influence of DOM
 - * Conflicting Results (Eriksson et al., 2004 vs. Leal et al., 2013)



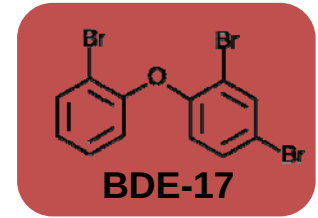
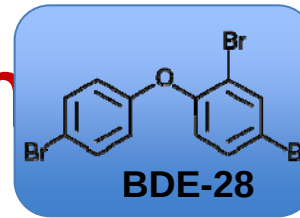
Methods



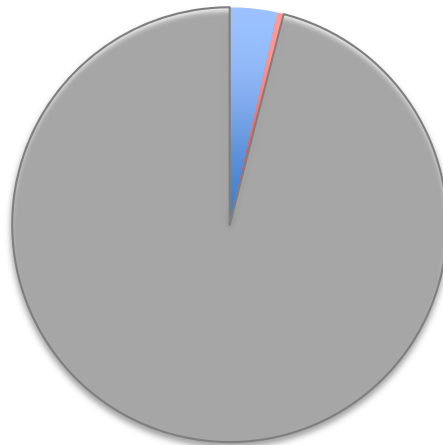
Results – MQ vs. DOM



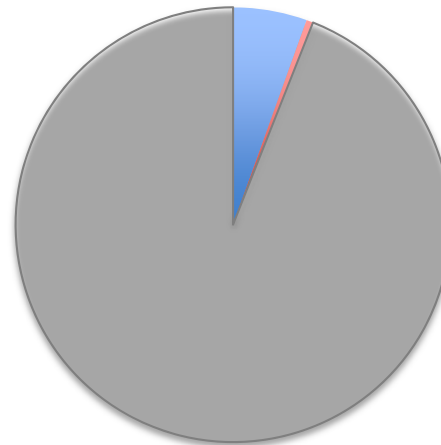
Product Form



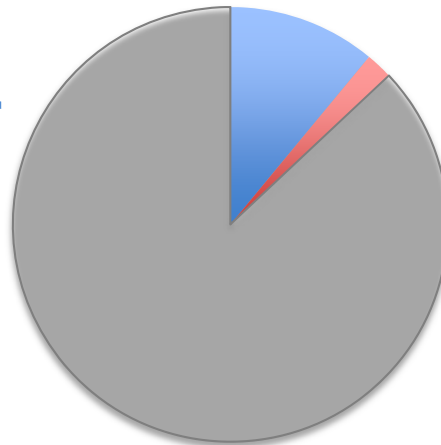
SRNOM



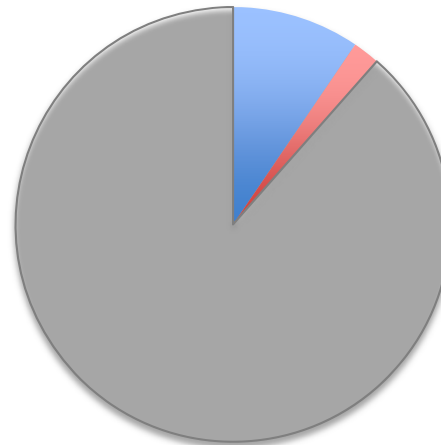
Sag Seep



PLFA

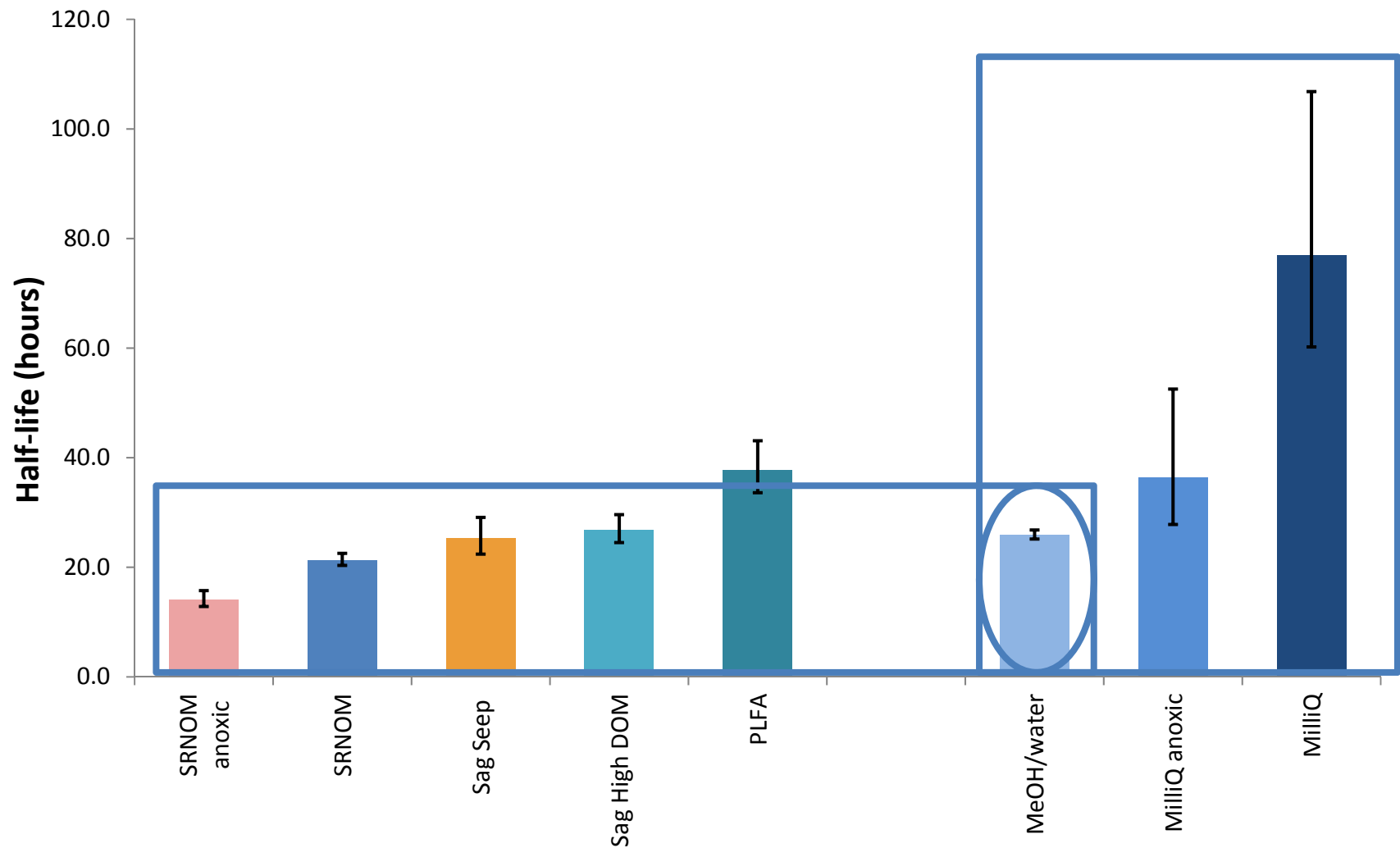


MilliQ

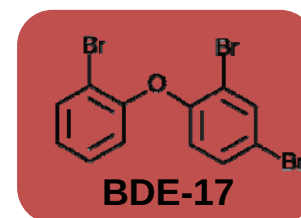
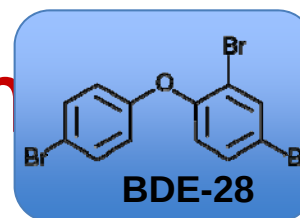


BDE-28
BDE-17
Product
Remainder

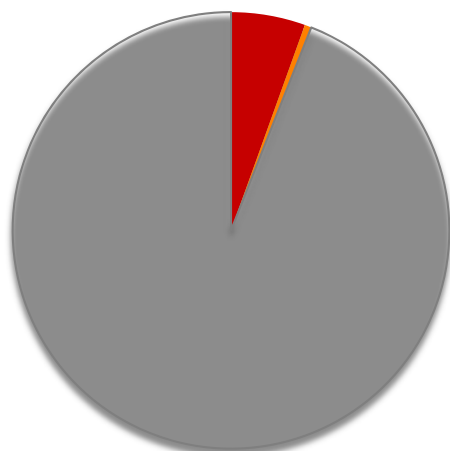
Organic Solvent Effects



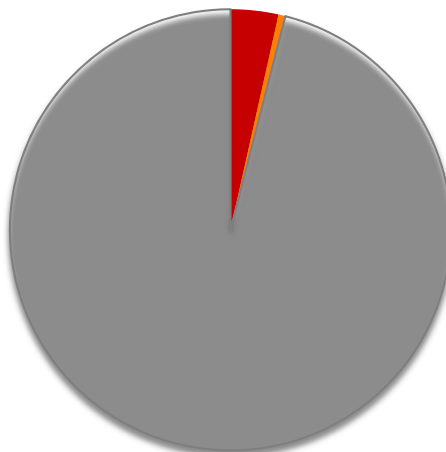
Product Form



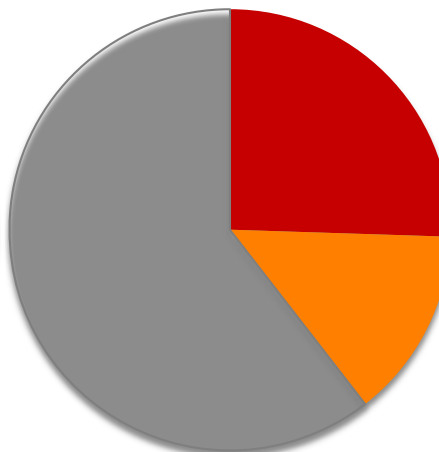
Sag Seep



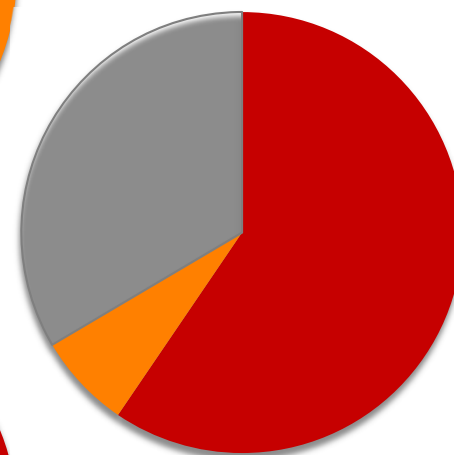
SRNOM



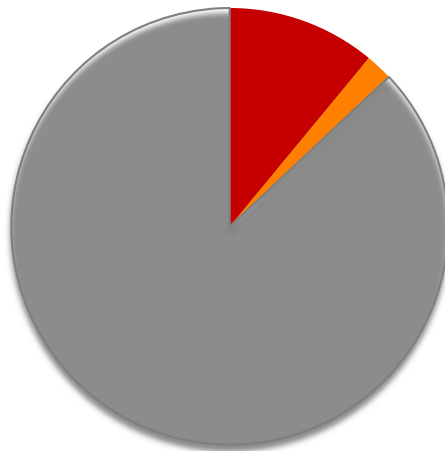
SRNOM anoxic



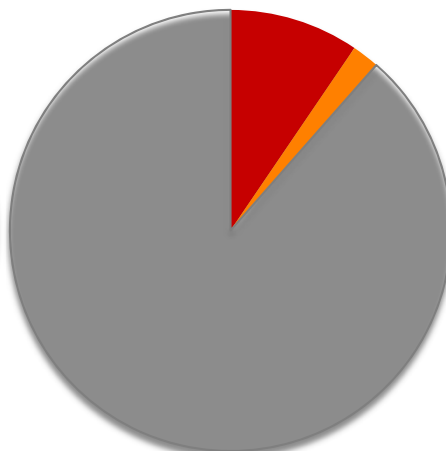
80/20 MeOH/Mq



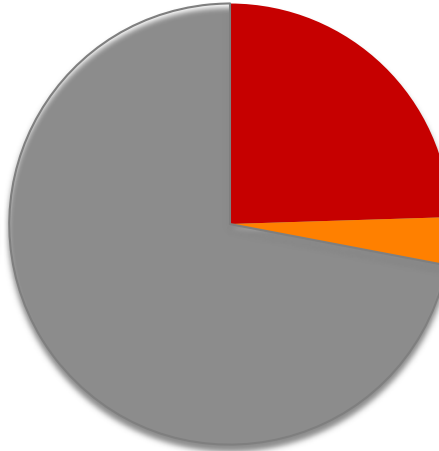
PLFA



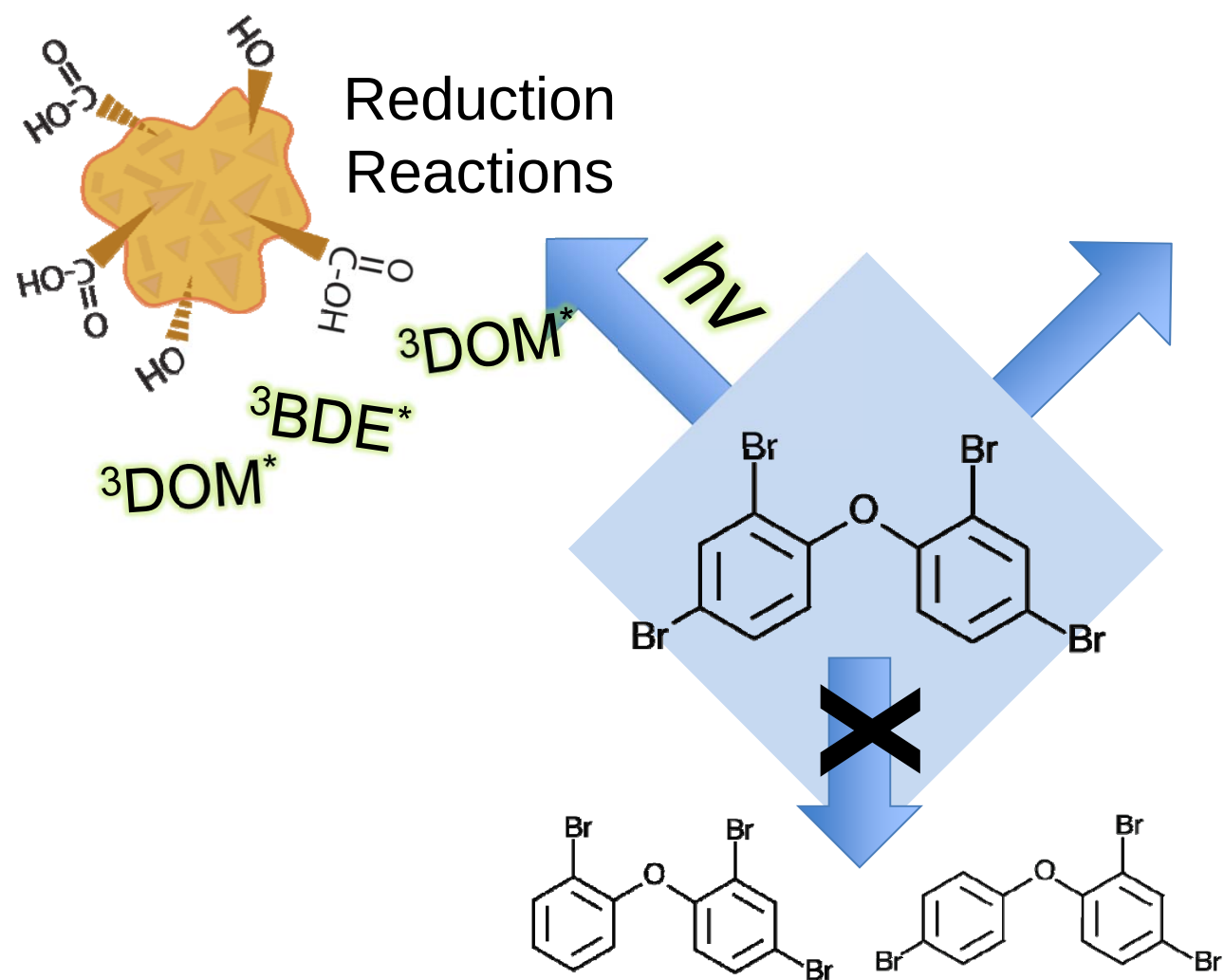
MilliQ



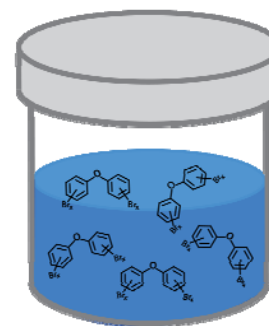
MQ anoxic



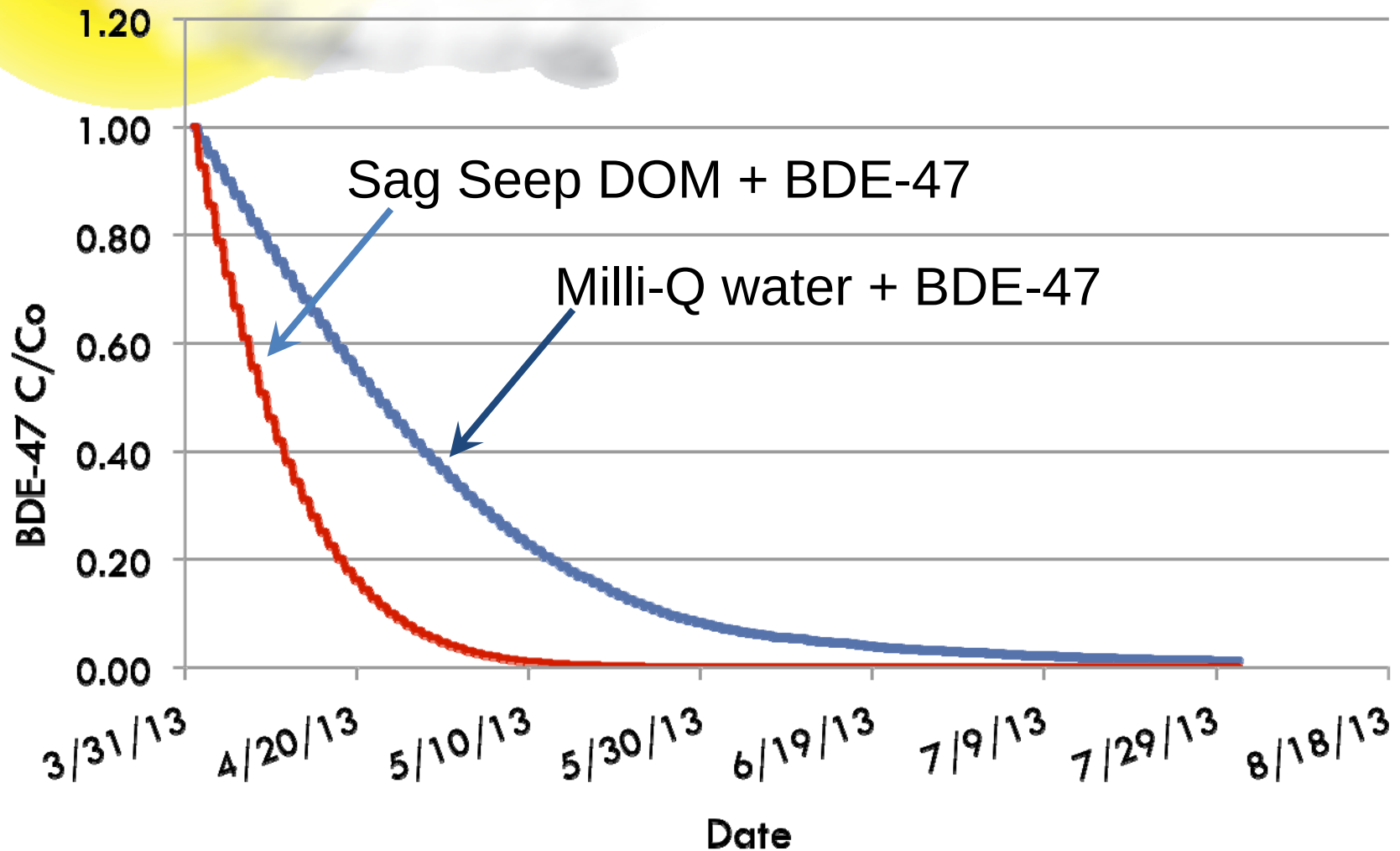
Conclusions 2



MeOH/Water



Arctic Photolysis model



Questions?



<http://www.alaska-in-pictures.com/bow-lake-sunrise-banff-national-park-8189-pictures.htm>